# Syntheses and Crystal Structures of Fused Thiophenes: [7]Helicene and Double Helicene, a $\mathbf{D}_{2}$-Symmetric Dimer of 3, $3^{\prime}$-Bis(dithieno $\left[2,3-b: 3^{\prime}, 2^{\prime}-d\right]$ thiophene) 

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The efficient synthesis of (TMS) $)_{2}$-[7]helicene (rac-3) and double helicene, a $\mathrm{D}_{2}$-symmetric dimer of $3,3^{\prime}$-bis(dithieno-[2,3-b:3', $2^{\prime}-d$ ]thiophene) (rac-4) was developed. The crystal structures of $\mathbf{3}$ and $\mathbf{4}$ show both strong intermolecular $\pi-\pi$ interactions and $S \cdots S$ interactions. UV/vis spectra reveal that both $\mathbf{3}$ and $\mathbf{4}$ show significant $\pi$-electron delocalization.

In carbohelicene chemistry, ortho-fused aromatic compounds show aesthetic and miraculous structures, which have attracted organic chemists for many years. Recently, $[n]$ helicenes display extraordinary optical and electronic properties in the field of organic functional materials ${ }^{1,2}$ and their synthetic methods have also been developed. ${ }^{3,4}$ As an important part of heterohelicenes,

[^0]annelated oligothiophenes have received more and more attentions recently, because such kind of materials provide significantly improved properties in organic electronics. ${ }^{5}$ Dithieno[2,3$\left.b: 3^{\prime}, 2^{\prime}-d\right]$ thiophene $(\mathbf{1})$, as one of the six possible isomeric dithienothiophenes has attracted considerable attention in preparation, ${ }^{6}$ has been used as the building blocks for higher homologues of helicenes. The series work on the synthesis of carbon-sulfur helicenes has been reported by Rajca, ${ }^{7}$ and a few reviews on the preparation of such helicenes are also published. ${ }^{7-10}$ The first and representative member in this family is tetrafunctionalized [7]helicene (2), which containing two Br and two TMS groups was reported in 2000. ${ }^{7 \mathrm{a}}$ Low yield (1.7\%) in making 2 with 3,4-dibromothiophene as starting material limits its application in materials science and the synthesis of higher homologues of helicenes.



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Herein, we describe the efficient synthesis of $(T M S)_{2}{ }^{-}$ [7]helicene (rac-3) and double helicene, a $\mathrm{D}_{2}$-symmetric dimer of $3,3^{\prime}$-bis(dithieno[2,3-b:3' $2^{\prime}-d$ ]thiophene) (rac-4), and their crystal structures. The synthetic route to them is shown in Scheme 1. The starting material is 2,5 -bis-trimethylsilanyl-dithieno[2,3-b:3', $2^{\prime}-d$ ]thiophene (5). The efficient synthesis of 5 (ca. $45 \%$ total yield with 3-bromothiophene as starting material) was reported in our previous work. ${ }^{6 \mathrm{~b}}$

Syntheses of 3 and 4. 5-Bromo-2-trimethylsilanyl-dithieno-[2,3-b:3 $\left.3^{\prime}, 2^{\prime}-d\right]$ thiophene (6) is obtained in $89-94 \%$ yield via the treatment of NBS directly with 5 in a mixture solvent ( $\mathrm{HOAc}-\mathrm{CHCl}_{3}$ ). At this step, one of two TMS groups in $\mathbf{5}$ is removed and bromination is acted in situ at same time. The next step is bromine dance reaction, ${ }^{60,12}$ which is the key step in the synthetic route to $\mathbf{3}$ and $\mathbf{4}$. The high efficient bromine

[^1]SCHEME 1. Synthetic Route to rac-3 and rac-4 ${ }^{a}$

${ }^{a}$ NBS $=N$-bromosuccinimide, LDA $=$ lithium diisopropylamide, TMS $=$ trimethylsilyl, TFA $=$ trifluoroacetic acid.
dance reaction is observed to $\mathbf{6}$ by using LDA in THF at $0{ }^{\circ} \mathrm{C}$ to generate 4-bromo-2-trimethylsilanyl-dithieno[2,3-b:3', $2^{\prime}$ d]thiophene (7) in $93-95 \%$ yield. Following the $\mathrm{Li} / \mathrm{Br}$ exchange on 7 , the resultant aryllithium species is oxidized with $\mathrm{CuCl}_{2}$ to afford 5,5'-di(trimethylsilanyl)-3,3'-bis-dithieno[2,3-b:3', $2^{\prime}$ d]thiophene (8) in 68-74\% yield. LDA-mediated lithiation is introduced to remove the protons on $\alpha$ positions in $\mathbf{8}$ to give dilithiated 8 . Dilithiated $\mathbf{8}$ is an amazing intermediate. On the one hand, the reaction of dilithiated $\mathbf{8}$ with bis(phenylsulfonyl)sulfide gives the annelated product [7]helicene: rac-3 in 56-61\% yield. On the other hand, dilithiated $\mathbf{8}$ could be oxidized with $\mathrm{CuCl}_{2}$ to generate rac-4 in $48 \%$ yield. The method in making 4 is similar to that of tetra[2,3-thienylene]s in Marsella's work ${ }^{13,14}$ and our previous work. ${ }^{6 b} 4$ is a novel $D_{2}$-symmetric dimer of $3,3^{\prime}$-bis(dithieno[2,3-b:3', $2^{\prime}-d$ ]thiophene), and a novel eight membered cyclic compound. With TFA to remove the TMS groups in rac-3, almost quantitive yield of rac-9 could be generated. Therefore, compounds rac- 3 and rac-4 are obtained in ca. $36 \%$ and ca. $29 \%$ isolated yields, respectively in four steps with $\mathbf{5}$ as starting material. From 3-bromothiophene as starting material, the overall yields for rac-3 and rac-4 could be ca. $16 \%$ and ca. $13 \%$ isolated yields, respectively in eight steps. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra show the expected 2 -fold symmetry for $\mathbf{3}, \mathbf{8}, \mathbf{9}$ and 4 -fold symmetry for $\mathbf{4}$ (see Supporting Information). 6 and $\mathbf{7}$ are isomers with significant difference in both ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra.

Crystal Analyses for 3 and 4. The structures of 7, 8, rac-3 and rac- $\mathbf{4}$ are all confirmed by Single-crystal X-ray analysis (Figure 1). In 7, all the thiophene rings are approximately coplanar. The $\mathrm{C} 10-\mathrm{C} 8-\mathrm{C} 6-\mathrm{C} 5$ torsion angle is $0.9^{\circ}$. In 8 , two moieties of dithieno[2,3-b:3', $2^{\prime}-d$ ]thiophene are nonplanar, the $\mathrm{C} 6-\mathrm{C} 5-\mathrm{C} 10-\mathrm{C} 9$ torsion is $50.6^{\circ}$ and the $\mathrm{H} 6-\mathrm{C} 6-\mathrm{C} 9-\mathrm{H} 9$ torsion is $65.6^{\circ}$, but each moiety is approximately coplanar with torsions of $2.6^{\circ}(\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4-\mathrm{C} 5)$ and $4.2^{\circ}(\mathrm{C} 10-\mathrm{C} 11-\mathrm{C} 12-\mathrm{C} 13)$, respectively. The dihedral angle between the two planes of moieties is $52.0^{\circ}$ and the distance between $\mathrm{H} 2 \mathrm{~A} \cdots \mathrm{H} 13 \mathrm{~A}$ is 3.25 Å.

With the formation of the middle thiophene ring from 8 to 3, the molecule of $\mathbf{3}$ is compressed with a helical structure (Figure 2B). The distance between $\mathrm{H} 2 \mathrm{~A} \cdots \mathrm{H} 15 \mathrm{~A}$ is changed to $2.36 \AA$ and the two H atoms are pointing away from each

[^2]

A


C


E

FIGURE 1. Molcular structure and conformation for 3, 4, 7 and 8. (A) top view for 7, (B) side view for 7, (C) top view for 8, (D) side view for $\mathbf{8}$, (E) top view for $\mathbf{3}$, (F) side view for $\mathbf{3}$ and (G) side view for 4.



B

FIGURE 2. $\pi-\pi$ interaction and $S-S$ interaction in crystal packing of rac-3. (A) crystal packing; (B) spacefill mode.
other (Figure 1). The repulsion of the facing terminal thiophene rings, causes an interplanar angle between the terminal thiophene rings, $43.0^{\circ}$, which is smaller than $54.1^{\circ}$ of $\mathbf{2},{ }^{7 \mathrm{a}}$ due to the steric hindrance decreased from the two Br atoms to two H atoms. The angles between the least-squares planes of neighboring thiophene rings are between $5.0^{\circ}$ and $9.6^{\circ}$. With the middle thiophene ring as a reference, the inner ( $\mathrm{C} 15, \mathrm{C} 14, \mathrm{C} 11, \mathrm{C} 10$, C7, C6, C3, C2) helix climbs $2.18 \AA$ and turns in-plane by $266^{\circ}$, and the analogous values for 2 are $2.92 \AA$ and $260^{\circ},^{7 a}$ respectively.

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FIGURE 3. $\pi-\pi$ interaction and $S-\pi$ interaction in crystal packing of macrocyclic dimer, rac-4. (A) crystal packing; (B) spacefill mode.

Crystal packing of rac-3 (Figure 2A) reveals exist of $\pi-\pi$ interaction between $(R)-\mathbf{3}$ and $(S)-\mathbf{3}$, the distances of $\mathrm{C} 8 \cdots \mathrm{C} 12$, $\mathrm{C} 9 \cdots \mathrm{C} 11, \mathrm{C} 9 \cdots \mathrm{C} 12, \mathrm{~S} 4 \cdots \mathrm{C} 13, \mathrm{~S} 5 \cdots \mathrm{C} 7$ and $\mathrm{S} 5 \cdots \mathrm{C} 10$ are $3.776,3.666,3.829,3.459,3.855$ and $3.751 \AA$, respectively. The interactions of S atoms include the interaction of neighboring enantiomers (the distance of $\mathrm{S} 1 \cdots \mathrm{~S} 5$ and $\mathrm{S} 1 \cdots \mathrm{~S} 6$ are 3.515 and $3.461 \AA$, respectively) and interaction between $(R)-3$ and $(S)-\mathbf{3}$ (the distance of $\mathrm{S} 3 \cdots \mathrm{~S} 6$ and $\mathrm{S} 4 \cdots \mathrm{~S} 6$ are 3.902 and 3.666 $\AA$, respectively).

The structure of 4 is a macrocyclic dimer of $3,3^{\prime}$-bis-(dithieno[2,3-b:3', $\left.2^{\prime}-d\right]$ thiophene) possessed approximately a $D_{2}$ point group of symmetry (Figure 1 ). ${ }^{15}$ The central cyclooctatetraene ring has a 'saddle' form with an average dihedral angle of $53.9^{\circ}$, smaller than that in tetra-o-phenylene derivatives ${ }^{15 a, 16}$ but bigger than that in tetra[2,3-thienylene]-(TMS) $4 .{ }^{6 \mathrm{~b}}$ In rac4, the sterically hindered biaryls correspond to bis(dithieno[2,3$\left.b: 3^{\prime}, 2^{\prime}-d\right]$ thiophene) moieties, and each of the moieties has a chiral axis with oppositive configurations. Each of the four dithieno[2,3-b:3' $\left.3^{\prime},-d\right]$ thiophenes is approximately coplanar, torsions are $2.66^{\circ}(\mathrm{C} 5-\mathrm{C} 6-\mathrm{C} 8-\mathrm{C} 10), 3.07^{\circ}(\mathrm{C} 13-\mathrm{C} 15-\mathrm{C} 16-\mathrm{C} 18)$, $7.05^{\circ}(\mathrm{C} 27-\mathrm{C} 28-\mathrm{C} 30-\mathrm{C} 32)$ and $3.49^{\circ}(\mathrm{C} 35-\mathrm{C} 36-\mathrm{C} 38-\mathrm{C} 39)$, respectively. The distances of $\mathrm{H} 5 \mathrm{~A} \cdots \mathrm{H} 27 \mathrm{~A}$ and $\mathrm{H} 18 \mathrm{~A} \cdots \mathrm{H} 39 \mathrm{~A}$ are 3.311 and $3.499 \AA$, respectively. Therefore, 4 shows somewhat of double-helix like configuration, in which two [7]helicene-like structures linked together pointing away from each other (Figure 1). On the other hand, two helix of the sulfurcontaining the edges of 4 embraced together to form the double helix (see spacefill mode in Figure 3B).
Crystal packing of rac-4 reveals two types of short contacts (Figure 3A). The first one is $\pi-\pi$ interaction between $(R)-4$ and (S)-4, the distances of $\mathrm{C} 14 \cdots \mathrm{C} 16, \mathrm{C} 14 \cdots \mathrm{C} 18, \mathrm{C} 15 \cdots \mathrm{C} 17$, $\mathrm{C} 28 \cdots \mathrm{C} 31, \mathrm{C} 29 \cdots \mathrm{C} 30$ are $3.778,3.719,3.619,3.775$ and 3.816 $\AA$, respectively. The second one is from the S atoms on terminal thiophene rings interact with the $\pi$-system of the neighboring 4. Two arms in $\mathbf{4}$ give two S atoms and another two arms provide two interacted $\pi$-systems. For example, the distances of S7 $\cdot$ S10, S7 $\cdots$ C34, S7 $\cdots$ C35, S7 $\cdots$ C36 and S7 $\cdots$ C37 are $3.575,3.552,3.466,3.333$ and $3.348 \AA$, respectively.

UV/Vis Spectra for 3 and 4. The UV/vis spectra for 3 and 4 in chloroform are shown in Figure 4. The integrated absorbance in $\mathbf{4}$ is approximately twice that of $\mathbf{3}$. $\mathbf{3}$ has $\pi$-electron delocalization including both helical distortion and possible conjugation through the sulfur atoms with maximum absorption peak at 267 nm .4 has four approximately planar dithieno[2,3-b:3', $2^{\prime}-d$ ]thiophenes, which are cross-conjugated together with two absorption peaks at $247,349 \mathrm{~nm}$ and a

[^3]

FIGURE 4. UV-vis spectra of $\mathbf{3}$ and $\mathbf{4}$ in chloroform at room temperature $\left([\mathrm{C}]=1 \times 10^{-5} \mathrm{M}\right)$.
shoulder peak at 354 nm . The bathochromic shift for $\mathbf{4}$ is rather substantial, due to its increased $\pi$-electron delocalization.

In summary, the synthesis of novel fused thiophenes: racemic carbon-sulfur helicene $\mathbf{3}$ and racemic $\mathrm{D}_{2}$-symmetric dimer 4 was efficiently developed. The crystal structures of 3 and 4 show both strong $\pi-\pi$ interactions and $\mathrm{S} \cdots \mathrm{S}$ interactions. Such short intermolecular contacts (especially, $\mathrm{S} \cdots \mathrm{S}$ ) are known to facilitate high carrier mobilities in organic semiconductors. ${ }^{7 d, 17-20}$ Therefore, $\mathbf{3}$ and $\mathbf{4}$ might be a novel compound with special properties in organic functional materials. Both module 3 and $\mathbf{4}$ are chiral, so the enantiomer of $\mathbf{3}$ could be used for making higher carbon-sulfur helicene. ${ }^{6 a}$ Both enantiomers of $\mathbf{3}$ and $\mathbf{4}$ might be used as chiral catalysts. ${ }^{21}$ The more work about $\mathbf{3}$ and $\mathbf{4}$ are in progress.

## Experimental Section

5-Bromo-2-trimethylsilanyl-dithieno[2,3-b:3', $\left.2^{\prime}-d\right]$ thiophene (6). $5(0.5000 \mathrm{~g}, 1.47 \mathrm{mmol})$ was dissolved in $30 \mathrm{~mL} \mathrm{CHCl}_{3}$, NBS ( $0.3135 \mathrm{~g}, 1.76 \mathrm{mmol}, 1.20$ equiv) was dissolved in 18 mL $\mathrm{HOAc}-\mathrm{CHCl}_{3}(\mathrm{v} / \mathrm{v}=4: 5)$ and added dropwise under good stirring at ambient temperature. After stirring for 2 h , the reaction mixture was quenched with 60 mL water and extracted with chloroform (3 $\times 15 \mathrm{~mL})$, then washed with saturated $\mathrm{NaHCO}_{3}(20 \mathrm{~mL})$ and water $(20 \mathrm{~mL})$. The organic layer was dried over anhydrous $\mathrm{MgSO}_{4}$. After the solvent was removed under vacuum, the residue was purified by recrystallization from $\mathrm{CHCl}_{3}-\mathrm{CH}_{3} \mathrm{OH}$ to yield $6(0.4550 \mathrm{~g}$, $89.1 \%$ ). From other two reactions on the 0.9826 and 1.9727 g scales of $\mathbf{5}, 0.9370 \mathrm{~g}(93.5 \%)$ and $1.8767 \mathrm{~g}(93.3 \%)$ of $\mathbf{6}$ were obtained, respectively. Mp $78-79{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.40$ (s, 1 H ), 7.38 (s, 1H), 0.37 (s, 9 H ), ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}, 100 \mathrm{MHz}$ ) $\delta 144.9,143.5,140.3,138.1,136.8,124.7,122.1,112.6,-0.1$. IR (KBr): 2952.6, 2923.8, 2889.4, $2840.4(\mathrm{C}-\mathrm{H}) \mathrm{cm}^{-1}$. MS (EI, 70 $\mathrm{eV}): m / z=345.89$ (42) $\left[\mathrm{M}^{+}\right], 330.86$ (100) $\left[\mathrm{M}^{+}-15\right]$.

4-Bromo-2-trimethylsilanyl-dithieno $\left[2,3-b: 3^{\prime}, 2^{\prime}-d\right]$ thiophene (7). $n-\mathrm{BuLi}$ ( 2.186 M in hexane, $0.11 \mathrm{~mL}, 0.24 \mathrm{mmol}, 1.5$ equiv) was added dropwise to diisopropylamine $(0.04 \mathrm{~mL}, 0.28 \mathrm{mmol}$, 1.8 equiv) in THF ( 5 mL ) at $0^{\circ} \mathrm{C}$. After 1 h at $0^{\circ} \mathrm{C}$, the prepared LDA solution was transferred by syringe into a solution of $\mathbf{6}(0.0551$ $\mathrm{g}, 0.16 \mathrm{mmol})$ in THF $(10 \mathrm{~mL})$ at $-78^{\circ} \mathrm{C}$. After 10 h at $-78^{\circ} \mathrm{C}$, methanol (excess) was added to quench the reaction. The reaction mixture was extracted with $\mathrm{CHCl}_{3}(3 \times 15 \mathrm{~mL})$ and washed with saturated $\mathrm{NaCl}(30 \mathrm{~mL})$ and water ( 30 mL ), and then dried over $\mathrm{MgSO}_{4}$. After the removal of the solvent under vacuum, the residue

[^4]was purified by column chromatography on silica gel with petrol ether $\left(60-90^{\circ} \mathrm{C}\right)$ as eluent to yield $7(0.0523 \mathrm{~g}, 94.9 \%)$ as a white solid. From other two reactions on the 0.3436 and 1.8317 g scales of $\mathbf{6}, 0.3223 \mathrm{~g}(93.8 \%)$ and $1.7079 \mathrm{~g}(93.2 \%)$ of 7 were obtained, respectively. Mp $79-81^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right): \delta 7.75$ (s, 1 H ), 7.28 (s, 1H), $0.39(\mathrm{~s}, 9 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{CDCl}_{3}, 100 \mathrm{MHz}\right)$ : $\delta 144.8,144.6,139.7,139.3,136.4,124.6,123.7,102.8,-0.1$. IR (KBr): 2952.6, 2923.8, 2892.8, $2855.1(\mathrm{C}-\mathrm{H}) \mathrm{cm}^{-1} . \mathrm{MS}(\mathrm{EI}, 70$ $\mathrm{eV}): m / z=345.88(40)\left[\mathrm{M}^{+}\right], 330.85(90)\left[\mathrm{M}^{+}-15\right]$.

5,5'-Di(trimethylsilanyl)-3,3'-bis-dithieno[2,3-b:3', $\left.\mathbf{2}^{\prime}-d\right]$ thiophene (8). To a solution of $7(0.0992 \mathrm{~g}, 0.29 \mathrm{mmol})$ in dry ethyl ether ( 10 mL ), $n$-BuLi ( $2.34 \mathrm{M}, 0.13 \mathrm{~mL}, 0.29 \mathrm{mmol}, 1.05$ equiv) was added dropwise at $-78{ }^{\circ} \mathrm{C}$. After keeping at $-78^{\circ} \mathrm{C}$ for 3 h , dry $\mathrm{CuCl}_{2}(0.1920 \mathrm{~g}, 1.43 \mathrm{mmol}, 5.00$ equiv) was added. The reaction mixture was kept at $-78^{\circ} \mathrm{C}$ for 3 h , then warmed up slowly to ambient temperature overnight. The reaction mixture was quenched with water, extracted with ethyl ether $(3 \times 20 \mathrm{~mL})$. The organic layer was washed with water $(3 \times 25 \mathrm{~mL})$ and then dried over $\mathrm{MgSO}_{4}$. The white product $8(0.0563 \mathrm{~g}, 74.3 \%)$ was obtained by column chromatography on silica gel with petrol ether (60-90 ${ }^{\circ} \mathrm{C}$ ) as eluent. From other reaction on the 0.6023 g scale of 7, 0.3137 $\mathrm{g}(67.6 \%)$ of $\mathbf{8}$ was obtained. Mp $183-185^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right.$, $400 \mathrm{MHz}): \delta 7.45$ (s, 2 H ), 6.79 (s, 2 H ), 0.14 (s, 18 H ), ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 100 \mathrm{MHz}\right): \delta 144.2,143.5,140.1,139.9,136.7,128.7$, 126.0, 124.8, -0.3. IR (KBr): 2950.7, $2890.9(\mathrm{C}-\mathrm{H}) \mathrm{cm}^{-1}$. HRMS (TOF MS EI ${ }^{+}$) $\mathrm{m} / \mathrm{z}$ calcd for $\left[\mathrm{C}_{22} \mathrm{H}_{22} \mathrm{Si}_{2} \mathrm{~S}_{6}\right] 533.9584$, found 533.9586.
[7]Helicene: (rac-3). $n$ - BuLi ( 2.51 M in hexane, $0.50 \mathrm{~mL}, 1.26$ $\mathrm{mmol}, 2.2$ equiv) was added dropwise to diisopropylamine ( 0.22 $\mathrm{mL}, 1.58 \mathrm{mmol}, 2.7$ equiv) in $\mathrm{Et}_{2} \mathrm{O}(15 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$. After 1.5 h at $0^{\circ} \mathrm{C}$, the prepared LDA solution was transferred by syringe into a solution of $\mathbf{8}(0.3071 \mathrm{~g}, 0.57 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(35 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$. After 3 h at $0{ }^{\circ} \mathrm{C}$, dry $\left(\mathrm{PhSO}_{2}\right)_{2} \mathrm{~S}(0.1894 \mathrm{~g}, 0.60 \mathrm{mmol}, 1.05$ equiv) was added at $-78{ }^{\circ} \mathrm{C}$, then the reaction mixture was kept at $-78{ }^{\circ} \mathrm{C}$ for 3 h and then warmed up slowly to ambient temperature overnight. The reaction mixture was quenched with water, extracted with ethyl ether $(3 \times 40 \mathrm{~mL})$ and then washed with saturated NaCl $(45 \mathrm{~mL})$ and water ( 45 mL ). After drying over $\mathrm{MgSO}_{4}$, the solvent was removed under vacuum. The residue was purified by column chromatography on silica gel with petrol ether $\left(60-90^{\circ} \mathrm{C}\right)$ as eluent to yield rac-3 ( $0.1993 \mathrm{~g}, 61.4 \%$ ) as a white solid. From other two reactions on the 0.3001 and 0.3028 g scales of $\mathbf{8}, 0.1849 \mathrm{~g}(58.4 \%)$ and $0.1780 \mathrm{~g}(55.6 \%)$ of rac- 3 were obtained, respectively. Mp
$259-261{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right): \delta 7.66(\mathrm{~s}, 2 \mathrm{H}), 0.25$ (s, 18 H$),{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 100 \mathrm{MHz}\right): \delta 145.0,143.5,141.1$, 140.1, 139.6, 130.4, 129.5, 128.1, 0.0. IR (KBr): 2952.6, 2892.8 $(\mathrm{C}-\mathrm{H}) \mathrm{cm}^{-1}$. HRMS (TOF MS EI ${ }^{+}$) $\mathrm{m} / \mathrm{z}$ calcd for $\left[\mathrm{C}_{22} \mathrm{H}_{20} \mathrm{Si}_{2} \mathrm{~S}_{7}\right]$ 563.9148, found 563.9151.
$\mathbf{D}_{2}$-Symmetric Dimer (rac-4). $n$-BuLi ( 2.64 M in hexane, 0.19 $\mathrm{mL}, 5.14 \mathrm{mmol}, 4.05$ equiv) was added dropwise to diisopropylamine ( $0.09 \mathrm{~mL}, 0.62 \mathrm{mmol}, 4.8$ equiv) in $\mathrm{Et}_{2} \mathrm{O}(6 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$. After 1 h at $0{ }^{\circ} \mathrm{C}$, the prepared LDA solution was transferred by syringe into a solution of $\mathbf{8}(0.0679 \mathrm{~g}, 0.13 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(15$ $\mathrm{mL})$ at $0{ }^{\circ} \mathrm{C}$. After 4.5 h at $0^{\circ} \mathrm{C}$, dry $\mathrm{CuCl}_{2}(0.17 \mathrm{~g}, 1.3 \mathrm{mmol}, 10$ equiv) was added at $-78^{\circ} \mathrm{C}$, then the reaction mixture was kept at $-78{ }^{\circ} \mathrm{C}$ for 4 h and warmed up slowly to ambient temperature overnight. The reaction mixture was quenched with water, extracted with ethyl ether $(3 \times 15 \mathrm{~mL})$. The organic layer was washed with saturated $\mathrm{NaCl}(25 \mathrm{~mL})$ and water $(2 \times 25 \mathrm{~mL})$, and then dried over $\mathrm{MgSO}_{4}$. The yellow product rac-4 was obtained by precipitation twice $\left(\mathrm{CHCl}_{3}-\mathrm{CH}_{3} \mathrm{OH}\right)(0.0327 \mathrm{~g}, 48.4 \%$ yield). From other reaction on the 0.1521 g of $\mathbf{8}, 0.0749 \mathrm{~g}(49.4 \%)$ of rac-4 was obtained. $\mathrm{Mp}>300{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR (benzene- $d_{6}, 400 \mathrm{MHz}$ ): $\delta 6.50$ $(\mathrm{s}, 4 \mathrm{H}),-0.08(\mathrm{~s}, 36 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 100 \mathrm{MHz}\right): \delta 143.6$, 143.0, 140.6, 137.3, 134.9, 134.8, 128.1, 126.5, -0.4. IR (KBr): 2953.2, 2920.4, 2893.2, $2850.5(\mathrm{C}-\mathrm{H}) \mathrm{cm}^{-1}$. HRMS (TOF MS $\mathrm{EI}^{+}$) $\mathrm{m} / \mathrm{z}$ calcd for $\left[\mathrm{C}_{44} \mathrm{H}_{40} \mathrm{Si}_{4} \mathrm{~S}_{12}\right]$ 1063.8856, found 1063.8848 .

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Supporting Information Available: Experimental details, NMR, HRMS spectra and X-ray crystallographic data. This material is available free of charge via the Internet at http://pubs.acs.org.

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