

Syntheses and Crystal Structures of Fused Thiophenes: [7]Helicene and Double Helicene, a D₂-Symmetric Dimer of 3,3'-Bis(dithieno[2,3-b:3',2'-d]thiophene)

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The efficient synthesis of $(TMS)_2$ -[7]helicene (rac-3) and double helicene, a D₂-symmetric dimer of 3,3'-bis(dithieno-[2,3-*b*:3',2'-*d*]thiophene) (rac-4) was developed. The crystal structures of 3 and 4 show both strong intermolecular π - π interactions and S····S interactions. UV/vis spectra reveal that both 3 and 4 show significant π -electron delocalization.

In carbohelicene chemistry, *ortho*-fused aromatic compounds show aesthetic and miraculous structures, which have attracted organic chemists for many years. Recently, [*n*]helicenes display extraordinary optical and electronic properties in the field of organic functional materials^{1,2} and their synthetic methods have also been developed.^{3,4} As an important part of heterohelicenes,

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annelated oligothiophenes have received more and more attentions recently, because such kind of materials provide significantly improved properties in organic electronics.⁵ Dithieno[2,3b:3',2'-d]thiophene (1), as one of the six possible isomeric dithienothiophenes has attracted considerable attention in preparation,⁶ has been used as the building blocks for higher homologues of helicenes. The series work on the synthesis of carbon–sulfur helicenes has been reported by Rajca,⁷ and a few reviews on the preparation of such helicenes are also published.^{7–10} The first and representative member in this family is tetrafunctionalized [7]helicene (2), which containing two Br and two TMS groups was reported in 2000.^{7a} Low yield (1.7%) in making 2 with 3,4-dibromothiophene as starting material limits its application in materials science and the synthesis of higher homologues of helicenes.



Herein, we describe the efficient synthesis of $(TMS)_{2}$ -[7]helicene (rac-3) and double helicene, a D₂-symmetric dimer of 3,3'-bis(dithieno[2,3-b:3',2'-d]thiophene) (rac-4), and their crystal structures. The synthetic route to them is shown in Scheme 1. The starting material is 2,5-bis-trimethylsilanyldithieno[2,3-b:3',2'-d]thiophene (5). The efficient synthesis of 5 (ca. 45% total yield with 3-bromothiophene as starting material) was reported in our previous work.^{6b}

Syntheses of 3 and 4. 5-Bromo-2-trimethylsilanyl-dithieno-[2,3-b:3',2'-d]thiophene (6) is obtained in 89–94% yield via the treatment of NBS directly with 5 in a mixture solvent (HOAc-CHCl₃). At this step, one of two TMS groups in 5 is removed and bromination is acted in situ at same time. The next step is bromine dance reaction,^{6a,12} which is the key step in the synthetic route to 3 and 4. The high efficient bromine

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SCHEME 1. Synthetic Route to rac-3 and rac- 4^{a}



^{*a*} NBS = *N*-bromosuccinimide, LDA = lithium diisopropylamide, TMS = trimethylsilyl, TFA = trifluoroacetic acid.

dance reaction is observed to 6 by using LDA in THF at 0 °C to generate 4-bromo-2-trimethylsilanyl-dithieno[2,3-b:3',2'd]thiophene (7) in 93–95% yield. Following the Li/Br exchange on 7, the resultant aryllithium species is oxidized with CuCl₂ to afford 5,5'-di(trimethylsilanyl)-3,3'-bis-dithieno[2,3-b:3',2'd]thiophene (8) in 68-74% yield. LDA-mediated lithiation is introduced to remove the protons on α positions in 8 to give dilithiated 8. Dilithiated 8 is an amazing intermediate. On the one hand, the reaction of dilithiated 8 with bis(phenylsulfonyl)sulfide gives the annelated product [7]helicene: rac-3 in 56–61% yield. On the other hand, dilithiated 8 could be oxidized with CuCl₂ to generate rac-4 in 48% yield. The method in making 4 is similar to that of tetra[2,3-thienylene]s in Marsella's work^{13,14} and our previous work.^{6b} **4** is a novel D₂-symmetric dimer of 3,3'-bis(dithieno[2,3-b:3',2'-d]thiophene), and a novel eight membered cyclic compound. With TFA to remove the TMS groups in rac-3, almost quantitive yield of rac-9 could be generated. Therefore, compounds rac-3 and rac-4 are obtained in ca. 36% and ca. 29% isolated yields, respectively in four steps with 5 as starting material. From 3-bromothiophene as starting material, the overall yields for rac-3 and rac-4 could be ca. 16% and ca. 13% isolated yields, respectively in eight steps. ¹H and ¹³C NMR spectra show the expected 2-fold symmetry for 3, 8, 9 and 4-fold symmetry for 4 (see Supporting Information). 6 and 7 are isomers with significant difference in both ¹H and ¹³C NMR spectra.

Crystal Analyses for 3 and 4. The structures of 7, 8, rac-3 and rac-4 are all confirmed by Single-crystal X-ray analysis (Figure 1). In 7, all the thiophene rings are approximately coplanar. The C10–C8–C6–C5 torsion angle is 0.9° . In 8, two moieties of dithieno[2,3-*b*:3',2'-*d*]thiophene are nonplanar, the C6–C5–C10–C9 torsion is 50.6° and the H6–C6–C9–H9 torsion is 65.6°, but each moiety is approximately coplanar with torsions of 2.6°(C2–C3–C4–C5) and 4.2°(C10–C11–C12–C13), respectively. The dihedral angle between the two planes of moieties is 52.0° and the distance between H2A····H13A is 3.25 Å.

With the formation of the middle thiophene ring from 8 to 3, the molecule of 3 is compressed with a helical structure (Figure 2B). The distance between H2A····H15A is changed to 2.36 Å and the two H atoms are pointing away from each



FIGURE 1. Molcular structure and conformation for **3**, **4**, **7** and **8**. (A) top view for **7**, (B) side view for **7**, (C) top view for **8**, (D) side view for **8**, (E) top view for **3**, (F) side view for **3** and (G) side view for **4**.



FIGURE 2. $\pi - \pi$ interaction and S-S interaction in crystal packing of rac-3. (A) crystal packing; (B) spacefill mode.

other (Figure 1). The repulsion of the facing terminal thiophene rings, causes an interplanar angle between the terminal thiophene rings, 43.0° , which is smaller than 54.1° of **2**,^{7a} due to the steric hindrance decreased from the two Br atoms to two H atoms. The angles between the least-squares planes of neighboring thiophene rings are between 5.0° and 9.6° . With the middle thiophene ring as a reference, the inner (C15, C14, C11, C10, C7, C6, C3, C2) helix climbs 2.18 Å and turns in-plane by 266° , and the analogous values for **2** are 2.92 Å and 260° ,^{7a} respectively.

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FIGURE 3. $\pi - \pi$ interaction and $S - \pi$ interaction in crystal packing of macrocyclic dimer, rac-4. (A) crystal packing; (B) spacefill mode.

Crystal packing of rac-3 (Figure 2A) reveals exist of $\pi - \pi$ interaction between (*R*)-3 and (*S*)-3, the distances of C8···C12, C9···C11, C9···C12, S4···C13, S5···C7 and S5···C10 are 3.776, 3.666, 3.829, 3.459, 3.855 and 3.751 Å, respectively. The interactions of S atoms include the interaction of neighboring enantiomers (the distance of S1···S5 and S1···S6 are 3.515 and 3.461 Å, respectively) and interaction between (*R*)-3 and (*S*)-3 (the distance of S3···S6 and S4···S6 are 3.902 and 3.666 Å, respectively).

The structure of 4 is a macrocyclic dimer of 3,3'-bis-(dithieno[2,3-b:3',2'-d]thiophene) possessed approximately a D_2 point group of symmetry (Figure 1).¹⁵ The central cyclooctatetraene ring has a 'saddle' form with an average dihedral angle of 53.9°, smaller than that in tetra-o-phenylene derivatives^{15a,16} but bigger than that in tetra[2,3-thienylene]-(TMS)₄.^{6b} In rac-4, the sterically hindered biaryls correspond to bis(dithieno[2,3b:3',2'-d thiophene) moieties, and each of the moieties has a chiral axis with oppositive configurations. Each of the four dithieno[2,3-b:3',2'-d]thiophenes is approximately coplanar, torsions are 2.66° (C5-C6-C8-C10), 3.07° (C13-C15-C16-C18), 7.05° (C27-C28-C30-C32) and 3.49° (C35-C36-C38-C39), respectively. The distances of H5A····H27A and H18A····H39A are 3.311 and 3.499 Å, respectively. Therefore, 4 shows somewhat of double-helix like configuration, in which two [7]helicene-like structures linked together pointing away from each other (Figure 1). On the other hand, two helix of the sulfurcontaining the edges of 4 embraced together to form the double helix (see spacefill mode in Figure 3B).

Crystal packing of rac-4 reveals two types of short contacts (Figure 3A). The first one is π - π interaction between (*R*)-4 and (*S*)-4, the distances of C14····C16, C14····C18, C15····C17, C28···C31, C29····C30 are 3.778, 3.719, 3.619, 3.775 and 3.816 Å, respectively. The second one is from the S atoms on terminal thiophene rings interact with the π -system of the neighboring 4. Two arms in 4 give two S atoms and another two arms provide two interacted π -systems. For example, the distances of S7····S10, S7····C34, S7····C35, S7····C36 and S7····C37 are 3.575, 3.552, 3.466, 3.333 and 3.348 Å, respectively.

UV/Vis Spectra for 3 and 4. The UV/vis spectra for 3 and 4 in chloroform are shown in Figure 4. The integrated absorbance in 4 is approximately twice that of 3. 3 has π -electron delocalization including both helical distortion and possible conjugation through the sulfur atoms with maximum absorption peak at 267 nm. 4 has four approximately planar dithieno[2,3-*b*:3',2'-*d*]thiophenes, which are cross-conjugated together with two absorption peaks at 247, 349 nm and a



FIGURE 4. UV-vis spectra of **3** and **4** in chloroform at room temperature ($[C] = 1 \times 10^{-5}$ M).

shoulder peak at 354 nm. The bathochromic shift for **4** is rather substantial, due to its increased π -electron delocalization.

In summary, the synthesis of novel fused thiophenes: racemic carbon–sulfur helicene **3** and racemic D₂-symmetric dimer **4** was efficiently developed. The crystal structures of **3** and **4** show both strong $\pi - \pi$ interactions and S…S interactions. Such short intermolecular contacts (especially, S…S) are known to facilitate high carrier mobilities in organic semiconductors.^{7d,17–20} Therefore, **3** and **4** might be a novel compound with special properties in organic functional materials. Both module **3** and **4** are chiral, so the enantiomer of **3** could be used for making higher carbon–sulfur helicene.^{6a} Both enantiomers of **3** and **4** might be used as chiral catalysts.²¹ The more work about **3** and **4** are in progress.

Experimental Section

5-Bromo-2-trimethylsilanyl-dithieno[2,3-b:3',2'-d]thiophene (6). 5 (0.5000 g, 1.47 mmol) was dissolved in 30 mL CHCl₃, NBS (0.3135 g, 1.76 mmol, 1.20 equiv) was dissolved in 18 mL $HOAc-CHCl_3$ (v/v = 4:5) and added dropwise under good stirring at ambient temperature. After stirring for 2 h, the reaction mixture was quenched with 60 mL water and extracted with chloroform (3 \times 15 mL), then washed with saturated NaHCO₃ (20 mL) and water (20 mL). The organic layer was dried over anhydrous MgSO₄. After the solvent was removed under vacuum, the residue was purified by recrystallization from CHCl₃-CH₃OH to yield 6 (0.4550 g, 89.1%). From other two reactions on the 0.9826 and 1.9727 g scales of 5, 0.9370 g (93.5%) and 1.8767 g (93.3%) of 6 were obtained, respectively. Mp 78-79 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.40 (s, 1 H), 7.38 (s, 1H), 0.37 (s, 9 H),¹³C NMR (CDCl₃, 100 MHz) δ 144.9, 143.5, 140.3, 138.1, 136.8, 124.7, 122.1, 112.6, -0.1. IR (KBr): 2952.6, 2923.8, 2889.4, 2840.4 (C-H) cm⁻¹. MS (EI, 70 eV): $m/z = 345.89 (42) [M^+]$, 330.86 (100) $[M^+ - 15]$.

4-Bromo-2-trimethylsilanyl-dithieno[2,3-b:3',2'-d]thiophene (7). *n*-BuLi (2.186 M in hexane, 0.11 mL, 0.24 mmol, 1.5 equiv) was added dropwise to diisopropylamine (0.04 mL, 0.28 mmol, 1.8 equiv) in THF (5 mL) at 0 °C. After 1 h at 0 °C, the prepared LDA solution was transferred by syringe into a solution of **6** (0.0551 g, 0.16 mmol) in THF (10 mL) at -78 °C. After 10 h at -78 °C, methanol (excess) was added to quench the reaction. The reaction mixture was extracted with CHCl₃ (3 × 15 mL) and washed with saturated NaCl (30 mL) and water (30 mL), and then dried over MgSO₄. After the removal of the solvent under vacuum, the residue

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was purified by column chromatography on silica gel with petrol ether (60–90 °C) as eluent to yield **7** (0.0523 g, 94.9%) as a white solid. From other two reactions on the 0.3436 and 1.8317 g scales of **6**, 0.3223 g (93.8%) and 1.7079 g (93.2%) of **7** were obtained, respectively. Mp 79–81 °C. ¹H NMR (CDCl₃, 400 MHz): δ 7.75 (s, 1 H), 7.28 (s, 1H), 0.39 (s, 9 H), ¹³C NMR (CDCl₃, 100 MHz): δ 144.8, 144.6, 139.7, 139.3, 136.4, 124.6, 123.7, 102.8, -0.1. IR (KBr): 2952.6, 2923.8, 2892.8, 2855.1 (C–H) cm⁻¹. MS (EI, 70 eV): m/z = 345.88 (40) [M⁺], 330.85 (90) [M⁺ – 15].

5,5'-Di(trimethylsilanyl)-3,3'-bis-dithieno[2,3-b:3',2'-d]thiophene (8). To a solution of 7 (0.0992 g, 0.29 mmol) in dry ethyl ether (10 mL), n-BuLi (2.34 M, 0.13 mL, 0.29 mmol, 1.05 equiv) was added dropwise at -78 °C. After keeping at -78 °C for 3 h, dry CuCl₂ (0.1920 g, 1.43 mmol, 5.00 equiv) was added. The reaction mixture was kept at -78 °C for 3 h, then warmed up slowly to ambient temperature overnight. The reaction mixture was quenched with water, extracted with ethyl ether $(3 \times 20 \text{ mL})$. The organic layer was washed with water $(3 \times 25 \text{ mL})$ and then dried over MgSO₄. The white product 8 (0.0563 g, 74.3%) was obtained by column chromatography on silica gel with petrol ether (60-90 °C) as eluent. From other reaction on the 0.6023 g scale of 7, 0.3137 g (67.6%) of 8 was obtained. Mp 183–185 °C. ¹H NMR (CDCl₃, 400 MHz): δ 7.45 (s, 2 H), 6.79 (s, 2 H), 0.14 (s, 18 H), ¹³C NMR (CDCl₃, 100 MHz): δ 144.2, 143.5, 140.1, 139.9, 136.7, 128.7, 126.0, 124.8, -0.3. IR (KBr): 2950.7, 2890.9 (C-H) cm⁻¹. HRMS (TOF MS EI⁺) m/z calcd for $[C_{22}H_{22}Si_2S_6]$ 533.9584, found 533.9586.

[7]Helicene: (rac-3). n-BuLi (2.51 M in hexane, 0.50 mL, 1.26 mmol, 2.2 equiv) was added dropwise to diisopropylamine (0.22 mL, 1.58 mmol, 2.7 equiv) in Et₂O (15 mL) at 0 °C. After 1.5 h at 0 °C, the prepared LDA solution was transferred by syringe into a solution of 8 (0.3071 g, 0.57 mmol) in Et₂O (35 mL) at 0 °C. After 3 h at 0 °C, dry (PhSO₂)₂S (0.1894 g, 0.60 mmol, 1.05 equiv) was added at -78 °C, then the reaction mixture was kept at -78 °C for 3 h and then warmed up slowly to ambient temperature overnight. The reaction mixture was quenched with water, extracted with ethyl ether $(3 \times 40 \text{ mL})$ and then washed with saturated NaCl (45 mL) and water (45 mL). After drying over MgSO₄, the solvent was removed under vacuum. The residue was purified by column chromatography on silica gel with petrol ether (60-90 °C) as eluent to yield rac-3 (0.1993 g, 61.4%) as a white solid. From other two reactions on the 0.3001 and 0.3028 g scales of 8, 0.1849 g (58.4%) and 0.1780 g (55.6%) of rac-3 were obtained, respectively. Mp 259–261 °C. ¹H NMR (CDCl₃, 400 MHz): δ 7.66 (s, 2 H), 0.25 (s, 18 H), ¹³C NMR (CDCl₃, 100 MHz): δ 145.0, 143.5, 141.1, 140.1, 139.6, 130.4, 129.5, 128.1, 0.0. IR (KBr): 2952.6, 2892.8 (C–H) cm⁻¹. HRMS (TOF MS EI⁺) *m*/*z* calcd for $[C_{22}H_{20}Si_2S_7]$ 563.9148, found 563.9151.

D2-Symmetric Dimer (rac-4). n-BuLi (2.64 M in hexane, 0.19 mL, 5.14 mmol, 4.05 equiv) was added dropwise to diisopropylamine (0.09 mL, 0.62 mmol, 4.8 equiv) in Et₂O (6 mL) at 0 °C. After 1 h at 0 °C, the prepared LDA solution was transferred by syringe into a solution of 8 (0.0679 g, 0.13 mmol) in Et_2O (15 mL) at 0 °C. After 4.5 h at 0 °C, dry CuCl₂ (0.17 g, 1.3 mmol, 10 equiv) was added at -78 °C, then the reaction mixture was kept at -78 °C for 4 h and warmed up slowly to ambient temperature overnight. The reaction mixture was quenched with water, extracted with ethyl ether $(3 \times 15 \text{ mL})$. The organic layer was washed with saturated NaCl (25 mL) and water (2 \times 25 mL), and then dried over MgSO₄. The yellow product rac-4 was obtained by precipitation twice (CHCl₃-CH₃OH) (0.0327 g, 48.4% yield). From other reaction on the 0.1521 g of 8, 0.0749 g (49.4%) of rac-4 was obtained. Mp > 300 °C. ¹H NMR (benzene- d_6 , 400 MHz): δ 6.50 (s, 4 H), -0.08 (s, 36 H), ¹³C NMR (CDCl₃, 100 MHz): δ 143.6, 143.0, 140.6, 137.3, 134.9, 134.8, 128.1, 126.5, -0.4. IR (KBr): 2953.2, 2920.4, 2893.2, 2850.5 (C-H) cm⁻¹. HRMS (TOF MS EI^+) *m/z* calcd for [C₄₄H₄₀Si₄S₁₂] 1063.8856, found 1063.8848.

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Supporting Information Available: Experimental details, NMR, HRMS spectra and X-ray crystallographic data. This material is available free of charge via the Internet at http://pubs.acs.org.

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